

Section 22 1 Earths Structure Answer Key Academic PDF

Section 22 1 Earth's Structure Answer Key

Introduction

Understanding Earth's structure is fundamental to the study of geology, geography, and earth sciences. The section titled "Section 22 1 Earth's Structure" typically covers the core concepts related to the layers that make up our planet, their properties, composition, and significance. An answer key for this section offers clarity on essential facts, helps in self-assessment, and enhances comprehension. This article provides an in-depth exploration of Earth's structure, elaborating on the key points, and serving as a comprehensive guide to the answers related to this topic.

Overview of Earth's Structure

The Layers of Earth

Earth is composed of several concentric layers, each differing in composition, physical state, and properties. These layers are broadly categorized into three main parts:

- The Crust
- The Mantle
- The Core

Understanding these layers is crucial for grasping Earth's internal processes, including tectonic movements, volcanic activity, and seismic phenomena.

The Earth's Crust

Composition and Characteristics

The Earth's crust is the outermost layer and is relatively thin compared to the other layers. It is composed mainly of:

- Silicate rocks: such as granite and basalt

- Minerals: including quartz, feldspar, mica, and olivine

The crust is divided into two types:

- Continental Crust: Thicker, less dense, composed mainly of granite
- Oceanic Crust: Thinner, denser, composed mainly of basalt

Importance of the Crust

- It forms the Earth's surface where landforms, oceans, and ecosystems exist.
- It is the layer where human activity takes place.
- It is involved in tectonic activities such as earthquakes and mountain formation.

The Mantle

Composition and Physical State

Beneath the crust lies the mantle, which extends to about 2,900 kilometers below the surface. It is composed mainly of silicate rocks rich in magnesium and iron.

- The mantle is semi-solid but behaves plastically over geological time scales.
- It contains minerals like olivine, pyroxene, and garnet.

Layers within the Mantle

The mantle can be subdivided into:

- Upper Mantle: Includes the lithosphere and asthenosphere
- Lower Mantle: More rigid, extends to the core boundary

Role of the Mantle

- It is involved in convection currents that drive plate tectonics.
- It influences volcanic activity and the formation of mountain ranges.

The Earth's Core

Composition and Characteristics

The core is the innermost layer of Earth, comprising about 15% of Earth's volume but accounting for roughly one-third of its mass.

- It consists mainly of iron and nickel.
- The core is divided into two parts:
 - Outer Core: Liquid layer approximately 2,200 kilometers thick
 - Inner Core: Solid sphere with a radius of about 1,220 kilometers

Importance of the Core

- The movement of the liquid outer core generates Earth's magnetic field.
- It influences seismic wave behavior, helping scientists understand Earth's internal structure.

Seismic Evidence of Earth's Structure

How Seismic Waves Reveal Earth's Layers

Seismic waves generated by earthquakes provide critical evidence of Earth's internal composition.

- Primary waves (P-waves): Can travel through solids and liquids; speed varies with material
- Secondary waves (S-waves): Travel only through solids; their absence in certain regions indicates liquid layers

Discontinuities and Boundaries

- The Mohorovi?i? discontinuity (Moho) marks the boundary between the crust and mantle.
- The Gutenberg discontinuity marks the boundary between the mantle and outer core.
- The Lehmann discontinuity marks the boundary between the outer and inner core.

Key Points to Remember (Answer Key Highlights)

- Earth's outermost layer is the crust, composed of solid rocks.

- Beneath the crust lies the mantle, a semi-solid layer with convection currents.
- The innermost part is the core, comprised mainly of iron and nickel, with a liquid outer core and a solid inner core.
- Seismic wave behavior helps determine the properties and boundaries of Earth's layers.
- The Moho is the crust-mantle boundary; the Gutenberg and Lehmann discontinuities define the core boundaries.

Common Questions and Answer Key Tips

Q1: What are the main layers of the Earth's structure?

A: The main layers are the crust, mantle, outer core, and inner core.

Q2: What evidence supports the existence of Earth's layers?

A: Seismic wave studies, especially how waves reflect and refract at different depths, support the layered model.

Q3: Why is the Earth's outer core liquid?

A: S-wave absence and seismic studies show that S-waves do not travel through the outer core, indicating it is liquid.

Q4: What generates Earth's magnetic field?

A: The movement of the liquid iron in the outer core creates Earth's magnetic field through the geodynamo process.

Significance of Understanding Earth's Structure

- Helps in predicting earthquakes and volcanic eruptions.
- Aids in locating mineral and energy resources.
- Essential for understanding Earth's geological history and evolution.
- Critical for engineering, construction, and environmental planning.

Conclusion

The section "Section 22 1 Earth's Structure" is foundational in earth sciences, providing insight into the complex internal composition of our planet. An answer key enhances comprehension by summarizing critical points, clarifying misconceptions, and guiding students through essential

concepts. Mastery of Earth's structure not only enriches scientific knowledge but also fosters appreciation for the dynamic processes shaping our planet. Whether for academic purposes or general curiosity, a thorough understanding of Earth's layers forms the basis for further exploration into geology, seismology, and planetary sciences.

Section 22 1 Earth's Structure Answer Key: A Comprehensive Review

Understanding the Earth's structure is fundamental in the fields of geology, geography, and environmental science. The chapter titled Section 22 1 Earth's Structure provides a detailed overview of the Earth's internal composition, layers, and the processes that shape its surface. This review aims to dissect the key concepts, clarify difficult ideas, and serve as a thorough guide to mastering the content covered in the answer key.

Introduction to Earth's Structure

The Earth's structure is composed of several concentric layers, each with unique physical and chemical properties. These layers influence everything from volcanic activity to tectonic movements, shaping the planet's surface and sustaining life.

- The Earth's layers are classified mainly into three broad categories:

- The Crust
- The Mantle
- The Core

Understanding these layers' composition, thickness, and behavior is crucial for grasping the Earth's dynamic nature.

The Crust: The Earth's Outer Layer

Characteristics of the Crust

- The crust is the Earth's outermost layer.
- It is solid and relatively thin compared to the other layers.
- Thickness varies:
 - Continental crust: approximately 35-70 km thick.
 - Oceanic crust: about 5-10 km thick.
- Composed mainly of silicate rocks rich in elements like oxygen, silicon, aluminum, iron, calcium,

sodium, potassium, and magnesium.

Types of Crust

- Continental Crust
 - Composed primarily of granitic rocks.
 - Thicker but less dense.
 - Forms the continents and landmasses.
- Oceanic Crust
 - Composed mainly of basaltic rocks.
 - Thinner but denser.
 - Forms the ocean floors.

Importance of the Crust

- Supports terrestrial life.
- Contains soil and minerals vital for agriculture and industry.
- The layer where most geological activities like earthquakes and volcanic eruptions occur.

The Mantle: The Thick Middle Layer

Characteristics of the Mantle

- Extends from the base of the crust (~35 km deep) to about 2,900 km beneath the surface.
- Constitutes about 84% of Earth's volume.
- Composed mainly of silicate rocks rich in magnesium and iron.
- Behavior: Solid rock that can flow slowly over geological time scales (viscoelastic behavior).

Layers Within the Mantle

- Lithosphere: The rigid, brittle outer part of the mantle and crust. It includes the crust and the

uppermost mantle.

- Asthenosphere: A semi-fluid, ductile layer beneath the lithosphere, allowing tectonic plates to move.

Role of the Mantle

- Drives plate tectonics through convection currents.
- Responsible for volcanic activity when magma from the mantle reaches the surface.
- Plays a key role in the Earth's thermal regulation.

The Core: The Innermost Layer

Characteristics of the Core

- Divided into two parts:
 - Outer Core
 - Inner Core
- Composed mainly of iron and nickel.
- The outer core is liquid; the inner core is solid.

Outer Core

- Extends from about 2,900 km to 5,150 km beneath the surface.
- Thickness: approximately 2,250 km.
- Responsible for Earth's magnetic field due to its liquid metal convection.

Inner Core

- Extends from 5,150 km to Earth's center (~6,371 km).
- Thickness: approximately 1,200 km.
- Solid due to immense pressure, despite high temperatures.

Significance of the Core

- Generates Earth's magnetic field, protecting the planet from solar wind.
- Influences geological phenomena such as earthquakes and volcanic eruptions.

Understanding Earth's Layers Through Seismic Waves

Seismology provides vital clues about Earth's internal structure.

- When earthquakes occur, seismic waves travel through Earth's layers.
- Variations in wave speed and behavior reveal the properties of each layer.
- The discontinuities such as the Mohorovi?i? discontinuity (Moho) mark boundaries between layers.

Seismic Evidence Supporting Layered Structure

- P-waves: Can travel through solids and liquids; their speed changes at boundaries.
- S-waves: Only travel through solids; their inability to pass through the outer core confirms its liquid nature.
- The shadow zones of seismic waves help locate the core's boundaries.

Plate Tectonics and Earth's Structure

Understanding Earth's internal structure is essential to grasp plate tectonics, the theory explaining the movement of Earth's lithospheric plates.

- The lithosphere (crust + upper mantle) is divided into tectonic plates.
- These plates float on the semi-fluid asthenosphere.
- Movement of plates causes earthquakes, mountain formation, and continental drift.

Types of Plate Boundaries

- Divergent Boundaries (e.g., Mid-Atlantic Ridge)
- Convergent Boundaries (e.g., Himalayas)
- Transform Boundaries (e.g., San Andreas Fault)

Application and Relevance of the Answer Key

The Section 22 1 Earth's Structure Answer Key provides essential solutions and explanations to

guide students through complex concepts.

- Clarifies misconceptions about Earth's layers.
- Offers step-by-step answers to typical questions.
- Reinforces understanding of seismic evidence and geological processes.
- Aids in exam preparation by highlighting key points and common pitfalls.

Summary of Key Points

- The Earth's structure comprises the crust, mantle, and core, each with distinct properties.
- The crust is solid and thin, supporting life and human activity.
- The mantle is semi-fluid and drives tectonic movements.
- The core is metallic, generating Earth's magnetic field.
- Seismic studies are critical for understanding internal layers.
- Plate tectonics explains surface phenomena based on internal structure.

Conclusion

Mastering the concepts in Section 22 1 Earth's Structure and understanding the answer key's detailed solutions are fundamental for students and enthusiasts aiming to deepen their knowledge of Earth's internal composition. Recognizing the interconnectedness of Earth's layers and their influence on surface phenomena fosters a comprehensive appreciation of the planet's dynamic nature.

By thoroughly exploring each layer's characteristics, significance, and the scientific evidence supporting their existence, learners can develop a solid foundation in Earth sciences. Such knowledge not only prepares students for exams but also encourages curiosity about Earth's processes, inspiring further exploration into geosciences.

Remember: The Earth's structure is a testament to the planet's complex and ever-changing nature. Grasping its layers and their interactions is key to understanding geological phenomena and the Earth's behavior over geological time scales.

QuestionAnswer What is covered in Section 22 1 Earth's Structure in the curriculum? Section 22 1 Earth's Structure typically covers the composition, layers, and physical features of the Earth, including the crust, mantle, core, and their properties. Why is understanding Earth's structure important? Understanding Earth's structure helps explain geological phenomena like earthquakes, volcanic activity, and plate tectonics, and is essential for natural resource exploration and environmental studies. What are the main layers of the Earth discussed in Section 22 1? The main

layers include the crust, mantle, outer core, and inner core, each with distinct properties and compositions. How does the answer key help students in mastering Section 22 1? The answer key provides correct responses to questions, aiding students in verifying their understanding and preparing effectively for exams. What are common questions asked in the Section 22 1 Earth's Structure assessment? Common questions include identifying the Earth's layers, explaining their characteristics, and understanding the processes occurring within each layer. How can students use the answer key to improve their learning? Students can compare their answers with the key to identify gaps in their knowledge, understand correct concepts, and practice more effectively. Are there diagrams included in the Section 22 1 Earth's Structure answer key? Yes, diagrams illustrating Earth's layers and features are often included to enhance understanding and provide visual aids for better learning. What are some tips for studying Section 22 1 Earth's Structure? Review key concepts, practice with past questions, use diagrams for visualization, and verify your answers with the answer key for better retention. Where can students find the official answer key for Section 22 1 Earth's Structure? The official answer key is usually available on educational websites, school portals, or provided by teachers as part of study materials.

Related keywords: geology, earth layers, crust, mantle, core, earth structure diagram, earth science, section 22.1, earth's interior, answer key

Extractive metallurgy of rare earths

extractive metallurgy of rare earths plays a critical role in the development of modern technologies, including electronics, renewable energy systems, and advanced manufacturing. As the demand for rare earth elements (REEs) continues to surge due to their unique magnetic, luminescent, and electrochemical properties, understanding the processes involved in their extraction and purification becomes increasingly important. This comprehensive guide explores the principles, methods, challenges, and advancements in the extractive metallurgy of rare earths, providing valuable insights for industry professionals, researchers, and students alike.

Introduction to Rare Earth Elements and Their Significance

Rare earth elements comprise a group of 17 chemically similar metallic elements, including the 15 lanthanides plus scandium and yttrium. Despite their name, many REEs are relatively abundant in the Earth's crust but are rarely found in concentrated, economically viable deposits. Their unique electronic configurations make them indispensable for high-tech applications such as magnets, catalysts, phosphors, and batteries.

Overview of Extractive Metallurgy of Rare Earths

Extractive metallurgy of rare earths involves a series of chemical and physical processes designed to isolate individual REEs from their mineral sources. The main objectives include:

- Mineral beneficiation: concentrating REE-bearing minerals from ore.
- Leaching: dissolving REEs from the mineral matrix.
- Separation and purification: isolating and purifying individual REEs.
- Refining: producing high-purity rare earth metals and compounds.

The entire process chain is complex, requiring specialized techniques to overcome challenges such as similar chemical behaviors among REEs and environmental concerns.

Sources and Mineralogy of Rare Earths

Understanding the mineralogy is crucial for selecting appropriate extraction techniques.

Common REE Minerals

- Monazite (Ce, La, Th, Nd, Sm, Gd, Dy, Er, Y): a phosphate mineral rich in light REEs.
- Xenotime (YPO₄): contains yttrium and heavy REEs.
- Bastnäsite (Ce, La, Nd, Y): another carbonate mineral.
- Other deposits include ion-adsorption clays, carbonatites, and weathered rare earth deposits.

Implications for Extraction

Different mineral types necessitate varied processing routes, with monazite and bastnäsite being the primary sources globally.

Mining and Beneficiation of REE Minerals

The initial step involves mining the ore and reducing its impurities.

Mining Techniques

- Open-pit mining
- Underground mining

Beneficiation Processes

- Crushing and grinding: liberates minerals.
- Gravity separation: separates based on density differences.
- Magnetic separation: exploits magnetic properties.
- Flotation: uses reagents to selectively float REE minerals.
- Heavy media separation: for coarse mineral separation.

Effective beneficiation enhances the concentration of REEs and reduces downstream processing costs.

Extraction Methods for Rare Earths

Extraction techniques are tailored based on mineralogy, deposit type, and economic considerations.

Acid Leaching

- Process: Treats mineral concentrates with acids (e.g., sulfuric, hydrochloric, or nitric acids).
- Advantages: Efficient at extracting REEs; scalable.
- Challenges: Acid consumption; handling of acidic waste.

Alkali Roasting and Leaching

- Process: Roasting the mineral with alkaline agents (e.g., sodium carbonate), followed by leaching.
- Application: Particularly effective for monazite.
- Advantages: Converts insoluble phosphates into soluble compounds.

Solvent Extraction

- Principle: Uses organic solvents to selectively extract individual REEs from aqueous solutions.
- Process flow:
 - Leach mineral to produce an REE-rich solution.
 - Use solvent extraction to separate REEs based on their chemical affinities.
 - Stripping and regeneration of solvents for reuse.
- Advantages: High selectivity; suitable for refining.

Ion Exchange

- Utilized for fine separation of REEs, especially when purity requirements are stringent.
- Employs ion-exchange resins to adsorb specific REEs.

Separation and Purification of Rare Earths

Due to the chemical similarities of REEs, separating individual elements is complex and requires multiple steps.

Solvent Extraction Techniques

- Commonly used in industry to achieve high purity.
- Involves counter-current extraction with selective organic solvents.

Chromatography

- Used for laboratory-scale separation of trace REEs.
- Techniques include ion-exchange chromatography and simulated moving bed systems.

Precipitation and Crystallization

- Final purification steps involve precipitation of specific REEs, followed by crystallization to obtain pure compounds.

Refinement of Rare Earth Metals

After separation, REEs are converted into metal forms or compounds.

Reduction Processes

- Electrolytic reduction: Produces metallic REEs.
- Thermal reduction: Uses magnesium or calcium to reduce REE oxides.

Production of Rare Earth Alloys and Magnets

- REEs like neodymium and praseodymium are alloyed with iron or boron to produce strong permanent magnets.
- These are vital for electric motors, wind turbines, and electronics.

Environmental Considerations and Challenges

Extractive metallurgy processes pose environmental challenges.

- Radioactive waste: Monazite contains thorium and uranium, requiring careful handling.
- Chemical waste: Acid and solvent use generate waste streams needing treatment.
- Energy consumption: High energy input, especially in reduction and electrolysis steps.
- Recycling and sustainability: Emphasis on recovering REEs from end-of-life products to reduce environmental impact.

Recent Advances and Future Outlook

Technological advancements aim to improve efficiency, reduce environmental impact, and lower costs.

- Bioleaching: Use of microorganisms to leach REEs.
- Hydrometallurgical processes: Improved solvent systems and membrane technologies.
- Direct extraction methods: Using ionic liquids or novel reagents.
- Recycling: Enhanced methods for recovering REEs from electronic waste.

The future of extractive metallurgy of rare earths depends on sustainable practices, innovative technologies, and diversified feedstock sources to meet global demand responsibly.

Conclusion

The extractive metallurgy of rare earths is a complex but vital field underpinning modern technological advancements. It involves a series of intricate processes?from mineral beneficiation to high-purity metal production?that require careful optimization to address economic, technical, and environmental challenges. As the demand for REEs continues to grow, ongoing research and technological innovations will be essential to develop more sustainable, efficient, and environmentally friendly extraction methods. Understanding these processes is key to securing a reliable supply of these critical materials for future generations.

Extractive metallurgy of rare earths is a critical field that underpins the development of modern technologies, from electronics and renewable energy to defense and medical applications. As global

demand for rare earth elements (REEs) continues to surge, understanding the intricacies of their extraction and processing becomes ever more vital. This comprehensive guide aims to demystify the complex processes involved in the extractive metallurgy of rare earths, highlighting key techniques, challenges, and innovations shaping the industry today.

Introduction to Rare Earth Elements and Their Importance

Rare earth elements comprise a group of 17 chemically similar metallic elements, including the fifteen lanthanides plus scandium and yttrium. Despite their name, rare earths are relatively abundant in the Earth's crust but are seldom found in concentrated and economically viable deposits. Their unique properties—such as high magnetic strength, luminescence, and catalytic activity—make them indispensable in modern technology.

Overview of the Extractive Metallurgy Process

The extractive metallurgy of rare earths involves a series of complex steps aimed at transforming mineral ores into pure, usable forms. The key stages include:

- Mining and mineral beneficiation
- Leaching
- Separation and purification
- Refinement into individual elements

Each stage presents unique challenges, often requiring specialized techniques and environmental considerations.

Mining and Mineral Beneficiation

Types of Rare Earth Ore Deposits

Rare earths are typically hosted in specific mineral deposits, such as:

- Monazite: A phosphate mineral rich in light rare earths
- Bastnäsite: A carbonate mineral with high rare earth content
- Xenotime: A phosphate mineral often containing yttrium and heavy rare earths

Extraction Techniques at the Mine

Most extraction begins with mining these mineral deposits, followed by beneficiation processes to

concentrate the valuable rare earths.

Beneficiation Processes

- Crushing and Grinding: To liberate mineral particles
- Gravity Separation: Using differences in specific gravity
- Magnetic Separation: Exploiting magnetic properties
- Froth Flotation: Utilizing differences in surface chemistry

These processes aim to produce a concentrate with higher rare earth content, reducing downstream processing costs.

Leaching and Initial Extraction

Purpose of Leaching

Leaching transforms mineral concentrates into soluble forms, allowing for separation from unwanted gangue materials.

Common Leaching Agents

- Acid Leaching: Using sulfuric acid or hydrochloric acid
- Alkali Leaching: Less common, used in specific cases
- Bioleaching: Emerging environmentally friendly methods employing microorganisms

Process Overview

- Roasting (if necessary): To convert minerals into more soluble forms
- Leaching: Dissolving the rare earths into solution
- Solid-Liquid Separation: Removing insoluble residues

Challenges and Considerations

- Managing acid consumption
- Controlling solution pH
- Handling radioactive or hazardous residues

Separation and Purification of Rare Earths

The Core Challenge

Due to their similar chemical properties, separating individual rare earth elements from each other is a complex and resource-intensive process.

Techniques for Separation

- Solvent Extraction

A predominant method involving the transfer of rare earth ions between aqueous and organic phases. Key features include:

- Selectivity based on complexation differences
- Use of specific extractants (e.g., di-2-ethylhexyl phosphoric acid - D2EHPA)
- Multiple stages for high purity

- Ion Exchange

Utilizes resin beads that selectively adsorb certain ions:

- Employed for refining and separating specific REEs
- Often used after solvent extraction to further purify

- Precipitation Methods

Involving the formation of insoluble compounds:

- Phosphates
- Carbonates
- Hydroxides

Used to isolate individual REEs or prepare intermediates.

Purification Strategies

- Repeated solvent extraction steps
- Crystallization techniques
- Membrane separation technologies (emerging)

Environmentally Friendly Alternatives

Recent innovations focus on reducing solvent use and toxic waste, such as bio-based extractants and membrane separation.

Refinement into Individual Elements

Final Purification

Achieving high purity (often >99.99%) involves:

- Repeated solvent extraction cycles
- Crystallization
- Zone refining techniques

Conversion to Commercial Forms

The purified rare earths are often:

- Oxides: Used in magnets, phosphors
- Metallic forms: For alloying and specialized applications
- Alloys and compounds: Tailored for specific industrial uses

Environmental and Sustainability Considerations

Extractive metallurgy of rare earths poses significant environmental challenges, including:

- Radioactive waste management (e.g., thorium in monazite)
- Acid mine drainage
- Generation of toxic organic solvents
- Land degradation

Innovations are ongoing to develop greener extraction methods, such as:

- Bioleaching
- Recycling of REEs from end-of-life products
- Closed-loop solvent extraction systems

Recent Advances and Future Directions

Innovative Technologies

- Membrane separations: Offering energy-efficient alternatives
- Deep eutectic solvents: As eco-friendly extractants
- Electrochemical methods: For direct extraction and separation

Recycling and Circular Economy

Increasing focus on recovering REEs from electronic waste and industrial by-products to reduce dependence on primary mining.

Policy and Industry Trends

Global initiatives aim to diversify supply chains, promote sustainable mining practices, and develop domestic sources, especially in light of geopolitical considerations.

Summary and Key Takeaways

The extractive metallurgy of rare earths is a sophisticated, multi-stage process that transforms mineral ores into essential components for cutting-edge technology. It involves:

- Careful mineral beneficiation to concentrate REEs
- Leaching techniques tailored to mineral type
- Complex separation methods like solvent extraction and ion exchange
- Final purification to high-grade standards

While the industry faces environmental and economic challenges, ongoing research and technological innovation promise more sustainable and efficient approaches. As demand for rare earths continues to grow, mastering their extraction remains vital for powering the future of technological advancement.

In conclusion, understanding the extractive metallurgy of rare earths is crucial for stakeholders across mining, processing, and end-use industries. With a focus on technological innovation and environmental stewardship, the field is poised to meet future demands responsibly and sustainably.

Question What are the primary methods used in the extractive metallurgy of rare earth elements? The primary methods include beneficiation (such as flotation and magnetic separation), solvent extraction, ion exchange, and hydrometallurgical processes like leaching, followed by purification and separation techniques to isolate individual rare earths. What challenges are associated with the extraction of rare earths from mineral ores? Challenges include the similar chemical properties of rare earth elements making separation difficult, the presence of radioactive or toxic impurities, the environmental impact of processing, and the economic viability of extracting low-grade ores. How does solvent extraction facilitate the separation of rare earth elements? Solvent extraction uses selective organic solvents to partition rare earth ions based on their differing affinities, allowing for efficient separation of individual elements from complex mixtures after initial leaching. What role does ion exchange play in the purification of rare earths? Ion exchange resins selectively adsorb specific rare earth ions from aqueous solutions, enabling high-purity separation and purification through controlled elution processes. What are recent advancements in the extractive metallurgy of rare earths aimed at improving sustainability? Recent advancements include developing greener leaching agents, recycling and reusing process solvents, improving energy efficiency, and exploring alternative extraction methods like bioleaching to reduce environmental impact.

Related keywords: rare earth elements, mineral processing, beneficiation, rare earth concentrates, solvent extraction, ion exchange, leaching, flotation, metallurgy processes, ore refining

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